

Samarium Triflate-Catalyzed Halogen-Promoted Friedel–Crafts Alkylation with Alkenes

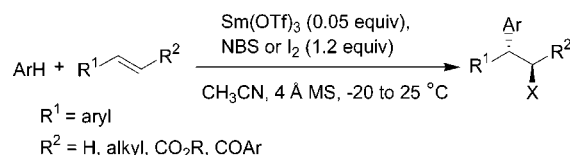
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ABSTRACT

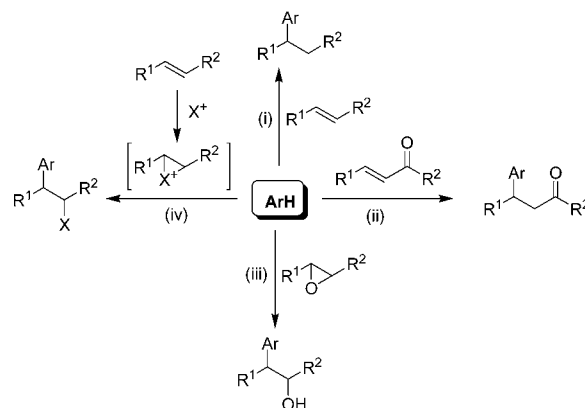


A versatile and efficient halogen-promoted highly regio- and stereoselective Friedel–Crafts (F-C) alkylation with alkenes has been developed with use of easily available and inexpensive NBS or I_2 as the efficient halogen sources. Lewis acids, in particular metal triflates, are found to be effective catalysts for this halogen-promoted F-C alkylation. Among these, Sm(OTf)_3 was the best catalyst. Electron-rich arenes smoothly underwent F-C alkylation with a variety of alkenes including α,β -unsaturated carbonyl compounds.

Friedel–Crafts (F-C) alkylation with alkenes is one of the most important C–C bond-forming reactions in organic chemistry.¹ These reactions are usually assisted by either protic acid or Lewis acid. On the basis of the mode of reaction, this F-C alkylation might be categorized as (i) reaction of arenes with normal alkenes via carbocation-like intermediates, (ii) reaction with electron-deficient alkenes via 1,4-conjugate addition, (iii) reaction of arenes with stable intermediate compounds derived from alkenes such as epoxides, and (iv) reaction with reactive intermediates generated in situ from alkenes such as three-membered halonium ions (Scheme 1). Among these, reactions with normal alkenes catalyzed by protic acid or Lewis acid have been extensively investigated. 1,4-Conjugate addition of arenes to electron-deficient alkenes are mostly limited to indoles as arenes.² Ring-opening of epoxides with arenes is also known in the literature.³ However, to the best of our knowledge, intermolecular F-C alkylation with alkenes via

halonium ions has not been reported. Interestingly, these alkylated arenes containing α -halo functionality could provide a further avenue for structural elaboration. Herein we report NBS/ I_2 -promoted intermolecular Friedel–Crafts alkylation with a variety of alkenes including α,β -unsaturated carbonyl compounds catalyzed by Sm(OTf)_3 that provide an alternate method for synthesis of 1,1-diaryl-2-halo alkanes and α,α -diaryl- β -halocarbonyl compounds, respectively.

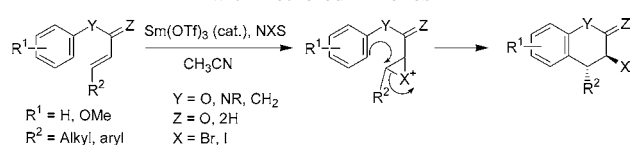
Scheme 1. Different Modes of F-C Alkylation with Alkenes



(1) For reviews of Friedel–Crafts alkylation reactions, see: (a) Olah, G. A. *Friedel–Crafts and Related Reactions*; Wiley-Interscience: New York, 1964; Vol. II, part 1. (b) Roberts, R. M.; Khalaf, A. A. *Friedel–Crafts Alkylation Chemistry: A Century of Discovery*; Dekker: New York, 1984. (c) Olah, G. A.; Krishnamurthi, R.; Prakash, G. K. S. *Friedel–Crafts Alkylations in Comprehensive Organic Synthesis*; Trost, B. M., Fleming, I., Eds.; Pergamon Press: Oxford, UK, 1991. (d) Bandini, M.; Melloni, A.; Umami-Ronchi, A. *Angew. Chem., Int. Ed.* **2004**, *43*, 550–556.

A current interest in our laboratory is the stereoselective 1,2-halo functionalization of alkenes.⁴ We found that Lewis acids, in particular metal triflates, activate NXS to facilitate the formation of halonium ions from alkenes. Recently we have reported Lewis acid-catalyzed and *N*-halosuccinimide (NXS)-promoted intramolecular F-C alkylation with tethered alkenes for the synthesis of annulated arene heterocycles and carbocycles (Scheme 2).^{4e} Accordingly, we anticipated that

Scheme 2. Halogen-Promoted Intramolecular F-C Alkylation with Tethered Alkenes



a suitable Lewis acid might, similarly, catalyze NXS-promoted intermolecular F-C alkylation of arenes with alkenes. For this purpose, we screened different Lewis acids, in particular metal triflates, as catalysts for halogen-promoted F-C alkylation of 1,2-dimethoxybenzene with 1,2-dihydronaphthalene (Table 1). $\text{Mg}(\text{OTf})_2$, $\text{La}(\text{OTf})_3$, $\text{Yb}(\text{OTf})_3$,

Table 1. Screening of Lewis Acids as Catalysts for the Halogen-Promoted F-C Alkylation with Alkenes

	ML_n	conv. ^a (%)	yield ^b of 3 (%)
1	none	20	<5 (12) ^c
2	CoCl_2	40	<5 (20) ^c
3	MgCl_2	65	8 (25) ^c
4	NiCl_2	80	<5 (35) ^c
5	CuCl_2	93	7 (35) ^c
6	Cu_2Cl_2	90	12 (27) ^c
7	$\text{Mg}(\text{OTf})_2$	100	41 (18) ^c
8	$\text{La}(\text{OTf})_3$	100	43 (21) ^c
9	$\text{Y}(\text{OTf})_3$	100	61 (17)
10	$\text{Yb}(\text{OTf})_3$	100	65 (18)
11	$\text{Zn}(\text{OTf})_2$	100	72 (12)
12	$\text{Cu}(\text{OTf})_2$	100	71 (10)
13	$\text{Sm}(\text{OTf})_3$	100	78 (7)

^a Determined from the ^1H NMR spectra of the crude reaction mixture with anthracene as internal standard. ^b Yields in parentheses refer to the yields of 1,2-dibromotetralin. ^c Along with 10–25% of nonseparable unidentified byproducts.

$\text{Y}(\text{OTf})_3$, $\text{Cu}(\text{OTf})_2$, $\text{Zn}(\text{OTf})_2$, and $\text{Sm}(\text{OTf})_3$ showed good catalytic activity (entries 7–13). Among these, $\text{Sm}(\text{OTf})_3$ was found to be the best catalyst and acetonitrile a suitable

solvent. It is noteworthy that in the absence of metal triflates, the reaction did not produce any F-C alkylated product (entry 1).

To study the scope of this halogen-promoted F-C alkylation, reactions of arenes with various types of alkenes were carried out (Table 2). In all cases, the reactions successfully

Table 2. $\text{Sm}(\text{OTf})_3$ -Catalyzed Halogen-Promoted F-C Alkylation with Alkenes^a

	alkene 2	ArH 1	temp (°C)	halogen source	time (h)	yield of 3 ^b (%)
1 ^c		1a	25	NBS	12	40
2 ^c		1b	25	NBS	12	43
3 ^d		1c	25	I_2	4	58
4 ^d		1d	25	I_2	3	52
5 ^d		1d	-20	NBS	0.5	51
6 ^d		1c	-20	NBS	0.5	62
7		1c	0	NBS	4	52
8		1d	0	NBS	3	75
9		1e	0	NBS	2	72 ^e

^a $\text{Sm}(\text{OTf})_3$ (0.05 equiv)-catalyzed F-C alkylation was performed with 1.0 equiv of alkene, 1.2 equiv of arene, 1.2 equiv of NBS (or 1.2 equiv of I_2 and 1.2 equiv of NaHCO_3) in CH_3CN . ^b Isolated yields of **3** after column chromatography; also 5–20% of halogen addition products were obtained. ^c Arene **1** was used as a cosolvent along with CH_3CN ($\text{CH}_3\text{CN}:\text{ArH}$ 4:1). ^d Alkene **2** was used in excess (2–3 equiv). ^e Along with 20% of trimethoxy bromobenzene as a nonseparable mixture.

proceeded to afford the corresponding alkylated products containing an α -halofunctionality in moderate to good yields. Toluene and *o*-xylene were used as cosolvents along with CH_3CN to afford clean products (entries 1 and 2). Electron-rich arenes such as anisole and 1,2-dimethoxybenzene underwent smooth reaction with styrene, when I_2 was used as a halogen source along with NaHCO_3 as an additive (entries 3 and 4). Halogen-promoted F-C alkylations with

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Table 3. Sm(OTf)₃-Catalyzed Halogen-Promoted F-C Alkylation with α,β -Unsaturated Carbonyl Compounds^a

entry	ArH 1	Ar'	R ³	time (h)	yield ^b (%)
1	1c	4-MeOC ₆ H ₄	OMe	12	77
2	1d	4-MeOC ₆ H ₄	OMe	12	91
3	1e	4-MeOC ₆ H ₄	OMe	12	80 ^c
4	1d	3,4-MeOC ₆ H ₃	OMe	12	85
5	1e	3,4-MeOC ₆ H ₃	OMe	12	76 ^c
6	1d	2-furyl	OMe	1	51
7	1d	3,4-MeOC ₆ H ₃	C ₆ H ₅	3.5	79
8	1e	3,4-MeOC ₆ H ₃	C ₆ H ₅	3.5	71 ^c
9	1d	4-MeOC ₆ H ₄	2-MeOC ₆ H ₄	4	90
10	1e	4-MeOC ₆ H ₄	2-MeOC ₆ H ₄	4	90 ^c

^a Sm(OTf)₃ (0.05 equiv)-catalyzed F-C alkylation was performed with 1.0 equiv of alkene, 1.2 equiv of arene, 1.2 equiv of NBS in CH₃CN at 15 °C. ^b Isolated yields of **5** after column chromatography. ^c 10–15% of trimethoxy bromobenzene was obtained.

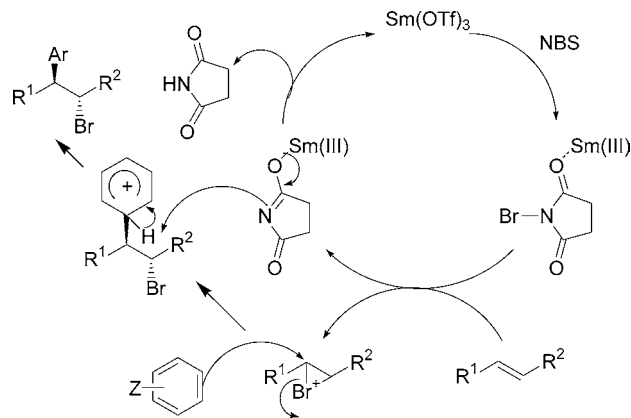
electron-rich alkenes, e.g., *p*-methoxy- β -methylstyrene and *p*-methoxy- β -propylstyrene, were carried out at –20 °C to afford clean F-C alkylated products. Improved yields were obtained when styrenes were used in excess (entries 5 and 6). Reactions of toluene and anisole with alkenes exclusively produced para-substituted adduct. No ortho-regioisomer was detected from the ¹H NMR of the crude reaction mixture. Reactions of anisole, 1,2-dimethoxybenzene, and 1,2,3-trimethoxybenzene with 1,2-dihydronaphthalene produced 1-aryl-2-halotetralins in good yields (entries 7–9). The other regioisomer 2-bromo-1-(3,4,5-trimethoxyphenyl)-1,2,3,4-tetrahydronaphthalene from 1,2,3-trimethoxybenzene was not detected. It is to be noted that 1-aryl-2-halotetralins could be key intermediates for the synthesis of many biologically important compounds⁵ by replacing the halogen atom with suitable nucleophiles.

α,β -Unsaturated carbonyl compounds represent a synthetically useful class of substrates for various alkene addition

reactions. In particular, halogen-promoted F-C alkylation of α,β -unsaturated carbonyl compounds, especially of cinnamate and chalcon would provide α -halo- β,β -diarylcarbonyl compounds, which could be transformed into various useful organic compounds by replacing the halogen with a series of nucleophiles. More interestingly, β,β -diarylcarbonyl compounds are important structural motifs found in many natural products and a number of pharmaceuticals.⁶ Different electron-rich arenes smoothly underwent NBS-promoted F-C alkylation with a number of cinnamates and chalcones at 15 °C and produced α -bromo- β,β -diarylcarbonyl compounds in high yields (Table 3).

Sm(OTf)₃ was found to be an efficient catalyst for the halogen-promoted F-C alkylation with alkenes including α,β -unsaturated carbonyl compounds and provided *anti*-1,2-haloalkylated products in moderate to good yields. It seems that Sm(OTf)₃ activates NBS by chelation to facilitate the formation of halonium ions from alkenes and the subsequent reactions with arenes led to the formation of F-C alkylated products along with regeneration of Sm(III) catalyst. Thus a plausible mechanism has been proposed for Sm(OTf)₃-catalyzed halogen-promoted F-C alkylation reactions (Scheme 3). However, further investigations are warranted to support this proposal.

Scheme 3. Proposed Mechanism for Sm(OTf)₃-Catalyzed Halogen-Promoted F-C Alkylation



We have developed a highly regio-, stereo-, and chemo-selective halogen-promoted Friedel–Crafts alkylation of aromatic compounds with alkenes using inexpensive and easily available NBS or I₂ as a halogen source. These reactions are catalyzed by Lewis acids, in particular metal triflates. Among these, Sm(OTf)₃ was found to be the best catalyst. Electron-rich arenes smoothly underwent chemo-selective halogen-promoted F-C alkylation with a variety of alkenes. This method was also found to be very effective with α,β -unsaturated carbonyl compounds such as

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cinnamates and chalcones and produced α -halo- β,β -diaryl-carbonyl compounds in high yields.

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Supporting Information Available: Detailed experimental procedures, and spectral data for all the compounds **3** and **5**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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